What's That Smell?

Purpose

Illustrate the variety of odors that are produced from esters.

Materials

- Organic alcohols in dropper bottles (see chart below for suggested alcohols)
- Organic acids in dropper bottles (see chart below for suggested acids)
- Sulfuric acid, H₂SO₄ (18 M)
- Sodium carbonate, Na₂CO₃
- Test tubes
- Test tube holder
- Bunsen burner

Safety

- Read the SDS sheets for all chemicals before using them.
- Wear safety glasses and gloves.
- Waft vapors when detecting odors.
- Concentrated acids are used.

Procedure

- 1. In a test tube, add 15 drops of ethanol to 20 drops of glacial acetic acid.
- 2. Add one drop of concentrated sulfuric acid.
- 3. Mix the reactants by gently tapping the test tube.
- 4. Gently heat the bottom of the test tube for several minutes. Keeping the test tube on a slight angle helps control vapor loss. If the solution starts to boil, remove it from the heat.
- 5. Add a small amount of sodium carbonate to neutralize any excess acid.
- 6. Carefully smell by wafting. Dilute with water if the smell is too strong.
- 7. See the chart for more combinations. Use 0.5 g of solid acids such as salicylic acid.

Results

• Vary depending on the organic acid and alcohol used (see reference chart below).

Follow-up Teaching Notes:

• Ester reference chart

Acid	Alcohol	Odor
acetic acid	ethanol	nail polish remover
acetic acid	1-butanol	raspberries
acetic acid	2-butanol	strawberries
acetic acid	1-pentanol	pear
acetic acid	isoamyl alcohol	banana
benzoic acid	ethanol	cherry
salicylic acid	methanol	Wintergreen
anthranilic acid	methanol	Grape
butanoic acid	ethanol	pineapple

• The concentrated sulfuric acid acts a catalyst and as a dehydrating agent for the condensation reaction between the alcohol and the carboxylic acid.

Ward's Science Tel: (866) 260-0501 • When using butyric acid, perform the demonstration in a fume hood due to the bad butyric acid odor.

Connections

• Organic synthesis, esters, condensation reactions.

Disposal/Clean-up

• Dispose of the esters down the drain with lots of water. Check local codes before any disposal activity.

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